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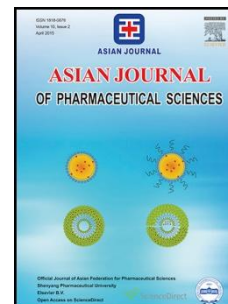
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Film Forming Systems for Topical and Transdermal Drug Delivery

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1 **Graphical abstract**

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3 **Film Forming Systems for Topical and Transdermal Drug Delivery**

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13 system changes significantly due to the loss of the volatile components of the vehicle which
14 results in formation of residual transparent film on the skin surface.

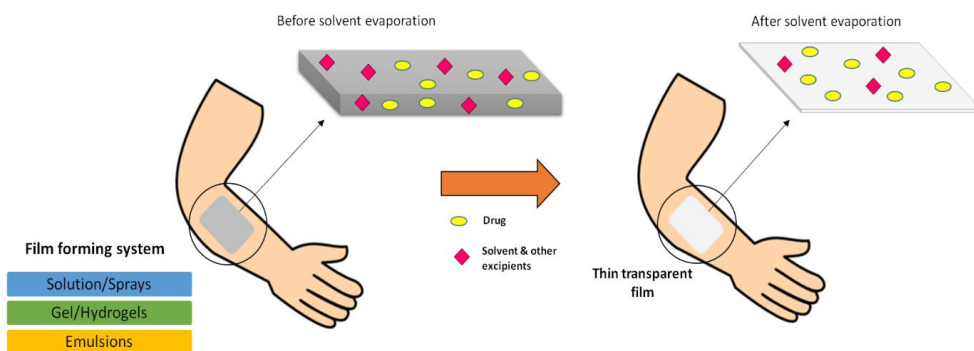
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Abstract:

22 Skin is considered as an important route of administration of drugs for both local and
 23 systemic effect. The effectiveness of topical therapy depends on the physicochemical
 24 properties of the drug and adherence of the patient to the treatment regimen as well as the
 25 system's ability to adhere to skin during the therapy so as to promote drug penetration
 26 through the skin barrier. Conventional formulations for topical and dermatological
 27 administration of drugs have certain limitations like poor adherence to skin, poor
 28 permeability and compromised patient compliance. For the treatment of diseases of body
 29 tissues and wounds, the drug has to be maintained at the site of treatment for an effective
 30 period of time. Topical film forming systems are such developing drug delivery systems
 31 meant for topical application to the skin, which adhere to the body, forming a thin transparent
 32 film and provide delivery of the active ingredients to the body tissue. These are intended for
 33 skin application as emollient or protective and for local action or transdermal penetration of
 34 medicament for systemic action. The transparency is an appreciable feature of this polymeric
 35 system which greatly influences the patient acceptance. In the current discussion, the film
 36 forming systems are described as a promising choice for topical and transdermal drug
 37 delivery. Further the various types of film forming systems (sprays/solutions, gels and
 38 emulsions) along with their evaluation parameters have also been reviewed.

39 **Keywords:** Film forming polymers; Topical drug delivery; Gelling agents.

40

41 1. Introduction

42 The skin is the most readily accessible organ of the body and acts as a barrier against the
43 micro and macromolecules of the environment because of its low permeability to such
44 substances [1]. Skin of an average adult body has approximately 2 m² surface area and it
45 receives about one-third of the total blood circulating throughout the body [2]. Percutaneous
46 absorption of drug through skin mainly occurs via stratum corneum. Stratum corneum is
47 made up of dead, keratinized epidermal cells having thickness of 10 µm and acts as a barrier
48 for permeation of drugs. Therefore transport of drug molecules across the skin is difficult [3].

49 The goal of drug administration through skin is for topical treatment of skin diseases or for
50 transdermal absorption of drugs in the systemic circulation. The topical route offers a large
51 and varied surface in addition to the ease of application via self-administration and provides
52 an alternative to oral delivery of drugs as well as hypodermic injection [4]. The rate and
53 extent of drug absorption through skin depends upon the skin physiology and
54 physicochemical properties of drugs as well as the delivery system. The current dosage forms
55 i.e. patches, ointments, creams, etc. are associated with several limitations. Patches have
56 various disadvantages, most commonly skin irritation [5], because of their occlusive
57 properties causing obstruction of sweat ducts, which in turn prevents loss of water vapor from
58 skin surface, difficulty in applying on the curved surfaces, pain while peeling off and poor
59 aesthetic appeal. Semisolid preparations like creams and ointments overcome some of these
60 drawbacks but have other limitations. These do not ensure persistent contact with the skin
61 surface and can be easily wiped off by patient's clothes [6]. Hence repeated application is
62 required in case of chronic diseases like athlete's foot, ringworm and candidiasis [7]. Also
63 these leave a sticky and greasy feel after application leading to poor patient compliance [8]
64 [9]. Therefore there is a need for development of a dosage form which permits less frequent
65 dosing by maintaining a close contact with the skin for prolonged time period thereby
66 improving the patient compliance.

67 Film forming system (FFS) is a novel approach which can be used as an alternative to
68 conventional topical and transdermal formulations. It is defined as non-solid dosage form that
69 produce a film in situ i.e. after application on the skin or any other body surface. These
70 systems contain the drug and film forming excipients in a vehicle which on contact with the
71 skin, leaves behind a film of excipients along with the drug upon solvent evaporation. The
72 formed film can either be a solid polymeric material that acts as matrix for sustained release

73 of drug to the skin or can be a residual liquid film which is rapidly absorbed in the stratum
74 corneum [10].

75

76 **2. Mechanism of film formation and permeation**

77 Film forming system is applied directly to the skin and it forms a thin, transparent film in situ
78 upon solvent evaporation as shown in the following Fig. 1.

79 After application of the formulation to the skin, the composition of the film forming system
80 changes significantly due to the loss of the volatile components of the vehicle which results
81 in formation of residual film on the skin surface. In this process the concentration of drug
82 increases reaching saturation level and with the possibility of reaching supersaturation level
83 on the skin surface. Supersaturation results in the enhanced drug flux through the skin by
84 increasing the thermodynamic activity of the formulation without affecting the skin's barrier,
85 thereby reducing the side effects or irritation. [10] [11].

86 The concept of supersaturation can be explained by modified form of Fick's law of diffusion.
87 Fick's law of diffusion is given by the equation 2.1:

$$88 \qquad \qquad \qquad J = \frac{DKC_v}{h} \qquad (2.1)$$

89 Where,

90 J= rate of drug permeation per unit area of skin per unit time (flux)

91 D= diffusion coefficient of drug

92 C_v= Concentration of drug

93 h= thickness of barrier to diffusion

94 From this equation, it is clear that the rate of drug permeation across skin is proportional to
95 the concentration of the drug. However this is true when all the drug is dissolved in the
96 vehicle.

97 Equation 2.2 describes modified form of Fick's law of diffusion:

$$98 \qquad \qquad \qquad J = \frac{\alpha D}{\gamma h} \qquad (2.2)$$

99 α= thermodynamic activity of drug within formulation

100 γ= thermodynamic activity of drug within membrane

101

102

103

104

105 According to this equation, the flux of the drug is directly proportional to the thermodynamic
106 activity of the system, which is related to saturation. However increasing the supersaturation
107 increases thermodynamic instability [12].

108 FFS creates supersaturated systems immediately after application to the skin, overcoming the
109 problem of instability. Thus improves the drug permeation through skin compared to other
110 transdermal dosage forms.

111 The delivering efficiency of the film forming solutions for ethinylestradiol was investigated.
112 The permeation of ethinylestradiol from the film forming solution prepared with enhancer or
113 without enhancer was compared to the permeation from the commercially available patch
114 (EVRA®) through human epidermis in vitro. The film forming formulations showed a higher
115 permeation than the commercial patch. Without enhancer the formulation transported more
116 than double the ethinylestradiol than the marketed patch. With enhancer, the formulation
117 delivered about seven times as much ethinylestradiol as that of the marketed patch. Thus
118 these systems prove to be useful in enhancing the drug permeation. [13]

119 **3. Comparison of Topical drug delivery systems**

120 FFS form an intermediate between the transdermal patches and semisolid dosage forms. Thus
121 exhibiting the advantages of both the systems. Table 1 summarizes the superiority of film
122 forming systems over patches and ointments. Fig. 2. depicts the drug permeation pattern of
123 all the three systems. In case of transdermal patches the drug is stored in a reservoir from
124 which the drug release occurs slowly and the drug is absorbed into the capillaries from where
125 it is transported to systemic circulation or it is formulated as a topical patch so as to penetrate
126 the skin to reach the target tissue for localized action. Drugs incorporated into semisolids
127 show their activity on the skin surface or penetrate into skin layers to reach the site of action
128 but systemic delivery of drugs is limited due to various factors. Film forming systems can
129 function as both semisolids and patches and can provide topical as well as transdermal
130 delivery as desired.

131 **4. Applications of film forming systems**

132 Initially film forming systems were predominantly used in the field of surgery or wound care.
133 Film forming solutions or gels have been used as tissue glues for the closing of operative
134 wounds. The film formers used for this purpose may be natural like fibrin or synthetic like

135 cyanoacrylates. These wound care preparations can be without drugs or with antimicrobial
136 agents to prevent infections in the wounds [14]. The film forming wound care products are
137 listed in table 2 [13]. It can also be used for non-medical uses, such as, the delivery of active
138 ingredients contained in beauty products like silicone film forming technologies used to
139 prepare cosmetic creams and ointments [15]. Transparent peel off masks technologies for
140 skin hydration, acne, etc. [16]. The film forming technology also has potential application as a
141 substrate for various barrier membranes used in industry. Barrier membranes are widely used
142 to protect workers from detergents, acids, bases and other hazardous chemicals, infra-red
143 heat, UV exposure etc. e.g. hydrophilic and hydrophobic creams and ointments, UV
144 protecting creams [17]. Film forming polymers are sprayed on the soil which forms a
145 membrane film increasing the integrity of soil and elevating the soil temperature, useful in
146 crop protection [18].

147

148

149

150 **5. Properties of film forming system**

151 The film forming preparation can be applied to the site regardless of shape and area, and can
152 be retained for a long time as compared to conventional semi-solid preparations. Fig. 3(A)
153 shows that FFS forms an almost completely transparent fast drying film on application. Fig.
154 3(B) shows that after drying a non-tacky, flexible and easily peelable film is formed. There is
155 an excellent adhesion of the formed film to the skin, hence wipe off resistance. Therefore the
156 risk of transfer of active ingredients to other people or clothes is reduced.

157

158 **6. Film forming Formulations**

159 *6.1. Sprays/Solutions:*

160 Film forming solutions and sprays is an attractive approach in transdermal dosage form. In
161 this the polymeric solution is applied to the skin as a liquid or sprayed on the skin and forms
162 an almost transparent film by solvent evaporation [19].

163 The film forming sprays/solutions are made up of four main components - drug, solvent
164 systems i.e. volatile and non-volatile vehicles, polymers and penetration enhancers. The non-
165 volatile component present in the solvent system prevents the drug from precipitating in
166 solution when the volatile solvent component evaporates. The non-volatile component is
167 chosen such that it itself partitions rapidly into the stratum corneum and also aids in
168 partitioning of the drug into the stratum corneum, as well as increases drug diffusivity by
169 disrupting the ordered intercellular lipids and enhance permeation. This type of delivery
170 system creates an invisible depot of drug in the stratum corneum from which the drug can be
171 slowly absorbed into the systemic circulation. Thus a sustained and enhanced permeation of
172 drug across the skin can be achieved following once a daily application [20] [21].

173 The formulation preparation involves addition of the polymer to the vehicle and stirring of
174 the solution overnight to ensure complete dissolution of the polymer. Once a clear polymeric
175 solution is obtained other optional excipients such as cross linker or plasticizer are added.
176 After addition of all excipients the solution is stirred for 24 h [22]. For the physical stability
177 of the API, the polymers are chosen such that they function as anti-nucleating agents and
178 crystallization inhibitors which prevent crystallization of drug even after solvent evaporation
179 e.g. polyvinyl pyrrolidone, polyethylene glycol, hydroxyl propyl methyl cellulose.

180 Film forming solutions can be applied with an applicator to the skin and allowed to dry. Film
181 forming spray is manufactured as a metered dose pump dispenser to provide fixed amount of
182 drug and it is sprayed on the topical site to form a film. These system forms a stable fast
183 drying, non-irritating invisible film from which the drug is available for transdermal therapy
184 [23]. Following administration, the film can be peeled off once the desired results are
185 obtained or for the termination of therapy as shown in Fig. 4.

186 Misra et al. prepared a liquid film forming solution using a mixture of polyvinyl pyrrolidone
187 and polyvinyl alcohol in isopropanol as film forming polymeric solutions for the biphasic
188 delivery of testosterone [24]. Ammar et al. developed a film forming polymeric solution of
189 ketorolac using eudragits and polyvinyl pyrrolidone in ethanol as film forming agents [22].
190 The mechanical properties and appearance of the prepared formulations were evaluated.

191 Mukesh *et al.* developed a fluconazole spray containing ethyl cellulose and eudragit RS 100
192 as film formers [25]. Yu ZW developed transdermal film-forming spray containing estradiol
193 and optimized the formulation using different polymers and plasticizers for efficient
194 penetration of estradiol for longer duration of time as compared to gel and patch [26].

195 6.2. *Gels:*

196 Gels are defined as semisolid dosage form containing both solid and liquid components. The
197 liquid component may be hydrophobic or hydrophilic in nature, immobilized in a three
198 dimensional network of the interconnected solid components [27]. Hydrogels are the aqueous
199 gels containing hydrophilic polymers that form three dimensional network in water [28].

200 The development of transdermal formulations is being focused on employing several
201 polymers as film forming agents along gelling agents. Various gelling agents are listed in the
202 table 3. The administration of film forming gel involves applying a dose on the arms,
203 shoulders, internal parts of the thighs or abdomen to form a thin bioadhesive film on the skin
204 [29]. The drug substance is dissolved in film forming vehicle and is thus incorporated in the
205 film formed on skin. The film can function as an external reservoir or limit the supply of drug
206 substance to the skin thereby controlling the release of drug [30].

207 Complete skin contact over the entire application is essential therefore the formulation
208 requires high flexibility to adapt to the movement of the skin, high substantivity, strong
209 adhesion to the skin for constant delivery and absorption of drug. Hence, along with gelling
210 agents, film forming agents, plasticizers, preservatives etc. are used in the formulation.
211 Compared to other forms, these systems offer easier use and application, appropriate
212 consistency and adhesiveness, good flexibility and elasticity and ease of manufacturing [31].

213 Saudagar et.al (2014) developed a film forming gel for prolonged delivery of Terbinafine
214 hydrochloride. The polymers used were eudragit and hydroxypropyl cellulose in combination
215 to provide a matrix film which would allow the release of terbinafine for a prolonged time.
216 The formulations were prepared using 3^2 full factorial design [30]. Xiang Li et.al developed a
217 film-forming gel formulation for sustained release of rotigotine with hydroxypropyl cellulose
218 and Carbomer 934. To optimize this formulation, the response surface analysis technique was
219 applied [32].

220
221 Film forming hydrogels are majorly used in wound healing. The formulation applied to the
222 wounded site provides a film that is resistant to physiological stress caused by the movement
223 of skin.

224 Liu et. al developed the formulation of sustained release transparent film-forming hydrogels
225 of tolterodine and studied the effects on stratum corneum with the help of response surface

226 methodology technique [33]. Dae-Duk Kim et.al developed a polyvinyl alcohol based soft
227 hydrogel of testosterone for transdermal delivery. The formulation was in semisolid phase
228 inside the tubes, but formed thin films within 2-3 min when applied to the skin [34].

229 6.3. Emulsions:

230 Emulsions are semisolid or liquid preparations having ability to solubilize both lipophilic and
231 hydrophilic drugs. Pharmaceutical emulsions consist of mixtures of aqueous phase and oily
232 phases stabilized by suitable emulsifying agents [38]. These can be oil-in water (o/w)
233 emulsions (oil phase is dispersed in the water phase) or water-in-oil (w/o) emulsions (water
234 phase dispersed in an oily continuous phase). The type of emulsion is formed depends mainly
235 on the type of emulsifiers, which is characterized by the hydrophilic-lipophilic balance
236 (HLB). The HLB is a scale from 1 to 20 and the higher the HLB, the more hydrophilic is the
237 surface active agent. An emulsifying agent is a substance which stabilizes the emulsion.
238 There are different types of emulsifying agents including surfactants, polymers, proteins
239 (gelatin) and finely divided solid particles (bentonite).

240 Film forming emulsions, in addition to the oil phase and the aqueous phase contain film
241 forming polymer. The volatile components present in the emulsions evaporate leading to the
242 changes in the tissue, allowing absorption of the drug [39]. The advantage of film forming
243 emulsions over semisolid formulations is that, it allows treatment of larger areas of affected
244 skin with an extended contact time and adequate substantivity, thus allowing sustained
245 dermal therapy of chronic diseases [40].

246 The delivery of the drug through skin depends on the nature of the API and the type of
247 emulsion. The dermal delivery of the lipophilic sunscreen agent ethylhexyl
248 methoxycinnamate was higher from the w/o emulsion than from the o/w emulsion most
249 probably because of the occlusive effect of the oily vehicle. But other studies have shown a
250 discrepancy. It was observed that the skin permeation of lipophilic parabens was enhanced
251 from o/w emulsions compared with the w/o emulsion. This was explained by a higher affinity
252 of the parabens for the vehicle than for the stratum corneum in case of the w/o emulsion [38].

253 Lunter et.al developed film forming emulsions for sustained dermal delivery of nonivamide
254 containing Eudragit NE and RS 30D as film formers [41]. In another study by Lunter, the *in*
255 *vitro* skin permeation and penetration of nonivamide from the prepared film-forming
256 emulsions was studied. It was found that the rate of permeation of the active substance is

257 determined by diffusion through the polymeric matrix in which the droplets were embedded.
258 Thereby, constant permeation rates and efficient API concentrations in the skin could be
259 maintained for a period of 12 h [40].

260

261 **7. Components of film forming systems**

262 *7.1. Drug*

263 For transdermal application of film forming systems, the drugs need to have suitable
264 properties which are independent of the dosage form. Generally the drugs which are
265 applicable to these systems are highly potent which permeate the skin rapidly, whose cause
266 no skin irritation and which are relatively stable to the enzymes present in the epidermis.
267 Other properties of the drug like partition coefficient, which dictates the pathway a drug will
268 follow through the skin. Secondly, molecular weight of drug is an important factor in drug
269 permeation as small molecules cross human skin than large molecules. The ideal properties of
270 the drug suitable of transdermal drug delivery system are listed in table 4.

271 *7.2 Polymers*

272 Polymers are the foundation of the FFS and variety of the polymers are available for the
273 preparation of these systems. In order to achieve the desired film properties, these polymers
274 can be used alone or in combination with other film forming polymers [42]. These polymers
275 should form a clear flexible film at skin temperature. The list of polymers along with their
276 molecular weight and properties are mentioned in table 5.

277 *7.3 Solvents*

278 The solvents form an important component in film formation. The solvent used in film
279 forming systems help in solubilizing the drugs as well as have an impact on drug permeation.
280 Commonly used solvents for topical and transdermal use [53] are listed in table 6. As these
281 solvents are widely used, the safety of these has been established on long term use.

282 *7.4 Plasticizers*

283 Plasticizers are used in the film forming systems to impart flexibility to the film and improve
284 the tensile strength of the film formed. The plasticizer used should be compatible with the

285 polymers used and should have low skin permeability. Commonly used plasticizers are
286 glycerine, polyethylene glycol, sorbitol, dibutyl phthalate, propylene glycol, triethyl citrate
287 etc. [54].

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8. Evaluation of film forming System

8.1. Film formation

The films are formed in a petri dish or on an excised pig ear skin. Film-formation is evaluated and rated as complete and uniform, incomplete or non-uniform, with or without precipitation of the film-forming polymer. The cosmetic aspects of the film are given in terms of transparency or opaque, sticky or dry, peelable or non-peelable [55]

8.2. Film flexibility

Film flexibility is evaluated on the basis of cracking and skin fixation and this is determined by stretching the skin in 2–3 directions. The film is rated flexible if there is no cracking or skin fixation and non-flexible if there is cracking and skin fixation.

8.3. Drying time

For the evaluation of the drying time the formulation is applied to the inner sides of the forearm of a volunteer. After a fixed time period a glass slide is placed on the film without pressure. If no liquid is visible on the glass slide after removal, the film is considered dry [56]. If remains of liquid are visible on the glass slide the experiment is repeated with an increase in drying time. A good FFS should have a minimum drying time to avoid long waiting time for the patient.

8.4. Stickiness

The stickiness of the film formed is determined by pressing cotton wool on the dry film with low pressure. Depending on the quantity of cotton fibres that are retained by the film the stickiness is rated high if there is dense accumulation of fibers on the film, medium if there is a thin fibre layer on the film and low if there is an occasional or no adherence of fibres. This evaluation

parameter is essential, as the formulation should be non-sticky to avoid adherence to the patients' clothes. [30]

8.5. Mechanical properties

The polymeric films are produced by solvent evaporation on a teflon plate. The dry films are cut with the help of a scalpel. Film thickness is measured with a digital micrometre. The mechanical properties of the films are determined with a tensile tester.

The tensile strength (σ) is calculated as:

$$\sigma = F_{\max} / A \text{ (N/m}^2\text{)}$$

Where F_{\max} (N) is the maximum force and A (m^2) is the cross-sectional area [19].

8.6. Determination of the water vapour permeability

The water vapour permeability is defined as the quantity of water transmitted through a unit area of film in unit time. This water vapour permeation data is important in determining the permeation characteristics of the film as it has influence on skin properties like hydration of stratum corneum, blood flow, and skin temperature [57]. Films are produced with a solvent evaporation technique on a teflon plate and dried for 72 h at room temperature. Circular samples are cut from the dry film sheets. For the sample preparation glass vials with an opening are filled with distilled water, covered with the circular film samples and a silicone ring and sealed tightly with an aluminium vial cap. The weight of the vial is determined and then placed into a desiccator creating an atmosphere of 58% relative humidity or low relative humidity (approximately 0%). They are kept at a determined temperature for 72 h and weighed after predetermined intervals. From the weight loss of the vials W (g) the water vapour permeability is calculated as the amount of water that permeates through the film in relation to the surface area A (cm^2) and the time t (h) [19]:

$$\text{WVP} = W/A * t$$

8.7. Swab studies

Swab test can be performed to evaluate the residence time of film forming system. For adhesion testing, glass was used as a polar, hydrophilic substrate. Glass was chosen as test surface because

films adhering strongly to it would also show strong adherence to skin because both materials display a polar surface structure [41].

Dry swab test: This test indicates the behaviour of FFS on the skin in dry condition. Dry swab test can be carried out on a glass plate. The glass plate is marked with 6 squares of 1x1 cm². Developed formulation is applied in this area. Dry cotton swabs of same volume are taken. Swabbing on the applied film is carried out at 0min, 30min, 2hrs, 4hrs, 6hrs and 8hrs and checked for drug content after extraction of drug from the swab.

Wet swab test: This test depicts the behaviour of FFS when it comes in contact with water or sweat. Procedure for the wet swab test is the same as dry swab test except the swab taken is soaked in water before and then the formulations are swabbed with this wet swab.

8.8. *Film Topography*

Atomic Force Microscopy (AFM) provides information about the topographic and mechanical properties of the polymeric films and helps to match the mechanical properties of the formed films to those of skin. It generates a nanoscale image of the film's homogeneity and roughness and requires no special treatment prior to the measurement [58].

8.9. *Film homogeneity*

Raman spectroscopy provides information about the chemical composition of the polymeric films. The chemical maps obtained from Raman spectra provides a measure of chemical homogeneity of films. Techniques based on Raman scattering can also be used to track the permeation of topically applied compounds through the skin [58].

8.10. *In vitro diffusion study*

The *in vitro* diffusion studies are used to predict the permeation characteristics of drug *in vivo*. Franz diffusion cell is used to determine the release profile of the drug from the film forming system. The cell is made up of two compartments, the donor and the receiver compartment between which the diffusion membrane is attached (egg membrane or cellophane). The donor compartment is exposed to atmosphere and the receptor compartment contains the diffusion medium. The sampling arm in the receptor compartment allows for sampling. Predetermined

quantity of the drug containing film forming formulation is placed on the donor compartment. Samples are collected and analyzed by suitable spectroscopic method for drug release [30].

8.11. Ex vivo permeation study

The ex vivo permeation studies are performed to study the effects of skin barrier on the developed film forming system. Franz diffusion cell/Keshary chein diffusion cell can be used for permeation study. Rat's skin is mounted between the two compartments, stratum corneum facing the donor compartment and dermis facing the receptor compartment. The formulation is applied to the skin surface which forms a film after drying. The receptor compartment contains phosphate buffer saline (pH 7.4) maintained at 37 ± 0.5 °C. Aliquots are collected at specific time intervals and analyzed by suitable spectroscopic method [59].

8.12. Skin penetration studies

The formulation is applied evenly on the skin using a pipette or a spatula. After fixed time intervals (e.g. 15 min, 1 h, 3 h, 6 h, 8 h, etc.) post application, the remaining formulation is removed. The film is wiped off with the help of cotton pads and the amount of drug present in the cotton pads is calculated, that is equivalent to the amount of drug remaining in the film. Therefore the amount of drug penetrated can be calculated by subtracting the remaining amount from the total amount of drug present in the formulation [60].

9. Commercialized film forming products

A number of companies have tried to develop film forming systems as a drug delivery platform and have marketed their products successfully. The companies with their products based on film forming technology are listed in table 7.

10. Conclusion and future prospects:

The film forming system presents a novel platform to deliver drugs to the skin both topical and transdermal. These film forming systems are simple and offer advantages of transparency, non-greasy, lower skin irritation, wipe off resistance, longer retention, greater increased dosage flexibility, improved patient compliance and aesthetic appearance Although considerable work

has been done on these systems, not much data is available on its delivery efficiency. Hence the marketed products available are less. Additional research is necessary to prove the relevance of film forming system as transdermal dosage form but the obtained results are encouraging for the further development of this novel topical drug delivering technology.

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Figure and Table Legends

Fig. 1: Mechanism of film formation

Fig. 2: Release profile of the topical and transdermal drug delivery systems

Fig. 3 Appearance of film forming system: (A) Formation of transparent film on application (B): Non-tacky, flexible, easily peelable film after drying

Fig. 4: Application of film forming solution on skin

Table 1: Comparison of topical drug delivery systems

Table 2: Film forming wound care products

Table 3: Gelling agents

Table 4: Ideal properties of drug for transdermal delivery

Table 5: Film forming polymers

Table 6: Examples of solvents used in transdermal system

Table 7: Commercialized film forming system

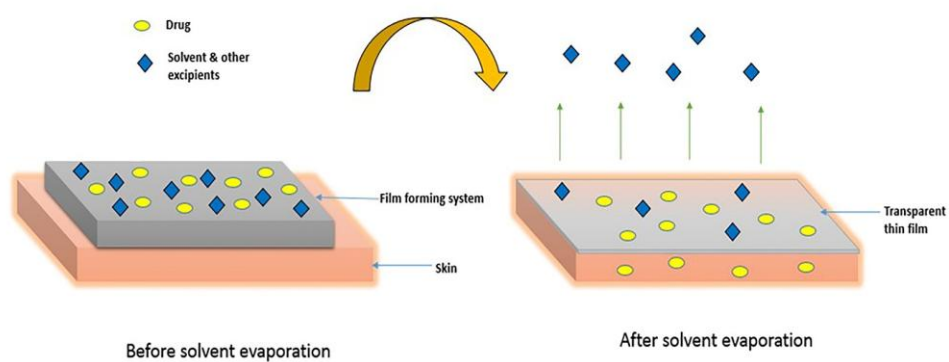


Fig. 1 Mechanism of film formation

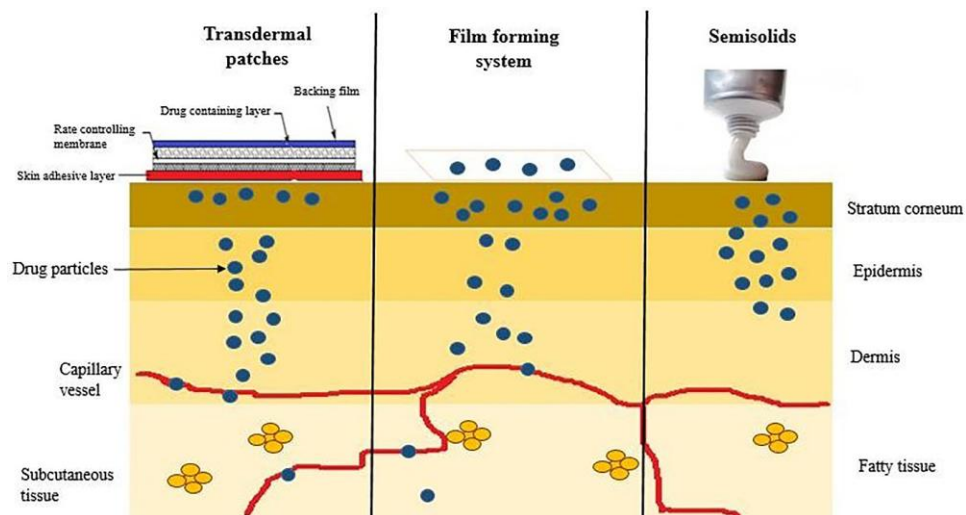


Fig. 2. Release profile of the topical and transdermal drug delivery systems

A



B



Fig. 3 Appearance of film forming system: (A) Formation of transparent film on application and (B) Non-tacky, flexible, easily peelable film after drying

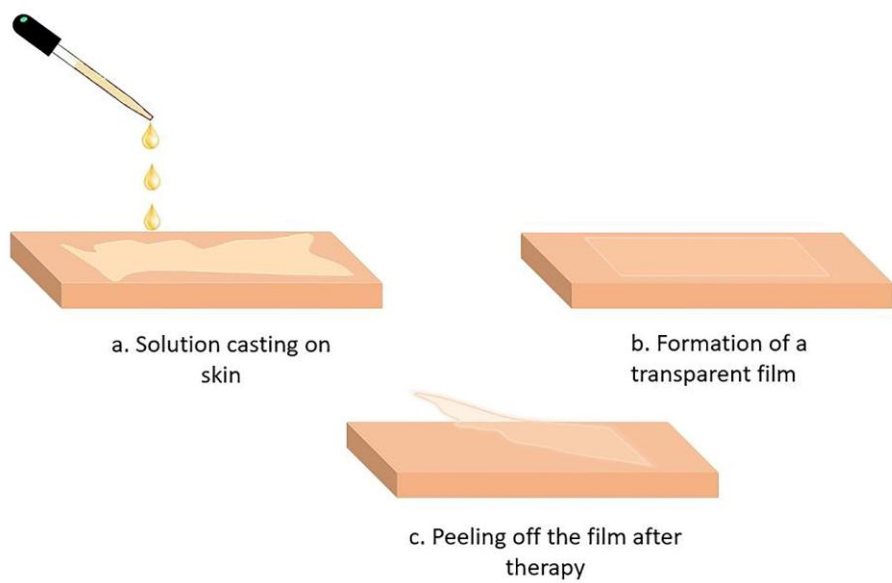


Fig. 4: Application of film forming solution on skin

Table 1 Comparison of topical drug delivery systems

	Patches	Film forming system	Semisolids
Visual Appearance	Highly Visible	Almost invisible	visible
Skin feel	Non-sticky, non-greasy	Non-sticky, non-greasy	Sometimes sticky, greasy
Administration	Convenient	Convenient	Sometimes messy
Dose adjustment	Low	High	High
Dosing frequency	1 – 7 days	1 – 2 days	1 day or less
Sustained release	Yes	Yes	No
Occlusive properties	Yes	No	No
Wipe off resistance	Yes	Yes	No
Residual remains	Possible	No	No

Table 2: Film forming wound care products [reproduced from ref. no. 14]

Trade names	Manufacturer	Film forming polymer
Dermabond [®]	Ethicon GmbH, Germany	Octylcyanoacrylate
EPIGLU [®] Gewebekleber	Meyer-Haake GmbH, Germany	Ethylcyanoacrylate, Poly(methylmethacrylate)
Flint [®] Sprühverband	Togal, Germany	Poly(butylmethacrylate, methylmethacrylate)
BandAid [®] Sprühpflaster	Ethicon GmbH, Germany	Cellulose Acetate Butanoate
Opsite [®] Spray	Smith & Nephew GmbH, Austria	Poly(methylacrylate)

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Table 3 Gelling agents

Gelling agent	Properties
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Gellan gum	<ul style="list-style-type: none">• Anionic polysaccharide• Gel properties depends on the presence of salts• Excellent biocompatibility and degrades into non toxic products
Carbomer (Carbopol®) Grades- ETD 2020, 171, 910, 934, 934P, 940, 1342 NF, 1971P [35]	<ul style="list-style-type: none">• Excellent gelling efficiency in low concentrations of 0.25% to 1.5%• pH dependent gelling• Forms clear gel in water
Carboxymethyl cellulose	<ul style="list-style-type: none">• Effective in concentrations of 4 to 6%.• Stable between pH 2 to 10• Incompatible with ethanol
HPMC (E series, F series, K4M, K15M, K100M)	<ul style="list-style-type: none">• Effective in concentrations of 2-6% as gelling agent.• Produces a smooth, clear spreadable gel.
Hydroxy ethyl cellulose	<ul style="list-style-type: none">• Effective in concentration of 0.5-1%• Has thickening and stabilizing properties.
Poloxamer (polyethylene- polypropylene glycol co- polymer) Grades- 124, 182, 188, 407 [36]	<ul style="list-style-type: none">• Used in the concentrations of 20-30 %• Thermoreversible gelation- liquid at refrigerated conditions and gel at room temperature.
Sepineo P 600 (acrylamide/sodium acryloyldimethyltaurate copolymer)	<ul style="list-style-type: none">• Concentration of 0.5-5% used as gelling agent• It has thickening properties in aqueous or organic media over a wide pH range. [37]

Table 4 Ideal properties of drug for transdermal delivery

Parameter	Properties
Dose	<10 mg/day
Half-life	10 h or less
Molecular weight	<500 Dalton
Partition coefficient Log P (octanol-water)	Between 1 and 3
Skin reaction	Non irritating and non-sensitizing
Oral bioavailability	Low

Table 5 Film forming polymers

Polymer	Properties
Hydroxypropyl	<ul style="list-style-type: none"> • Produce a light, non-greasy uniform film with good texture
Methylcellulose (HPMC)	<ul style="list-style-type: none"> • Do not interact significantly with other ingredients.
HPMC (E4M, E15, E50M K4M,) [43]	<ul style="list-style-type: none"> • Surface active agent, therefore adsorb water providing easy dispersion, lubricity and comfort feel in occlusive state on application to skin. [44]
Ethyl cellulose (EC)	<ul style="list-style-type: none"> • Nontoxic, nonirritating nonallergic material • Good film forming properties that form tougher films. [45]
Hydroxypropyl cellulose	<ul style="list-style-type: none"> • Nonionic, pH insensitive polymer • Water soluble [46]
Poly vinyl Pyrrolidone (PVP) (PVP K30, PVP VA64)	<ul style="list-style-type: none"> • Solubility in water and other solvents • Adhesive and binding property • Acts as a bioavailability enhancer [47]

Poly vinyl alcohol (PVA)	<ul style="list-style-type: none"> • Water soluble • Excellent film forming and adhesive properties • Nontoxic and biocompatible [48]
Chitosan	<ul style="list-style-type: none"> • Excellent film forming ability • Opens the tight junctions of mucosal membrane, thereby enhancing the paracellular permeability and penetration of drug. [49] • Controls drug release. [50]
Eudragit (polymethacrylates copolymer) Eudragit RS 100, RL 100, NE, RS 30D, S 100	<ul style="list-style-type: none"> • Transparent, elastic, self-adhesive • Good adhesion to the skin [51]
Silicones Polydimethylsiloxane (PDMS)	<ul style="list-style-type: none"> • Water vapor permeable film • Adequate substantivity and durable film [52]
Acrylates copolymer Avalure® AC 118, AC 120	<ul style="list-style-type: none"> • Tough, breathable, abrasion resistant films [14]

Table 6 Solvents used in topical systems

Category	Examples
Glycols	Propylene glycols, Polyethylene glycols
Alcohols	Ethanol, Butanol, Isopropanol, Benzyl alcohol, Lanolin alcohols, Fatty alcohols
Other solvents	Ethyl acetate, Oleic acid, Isopropyl myristate

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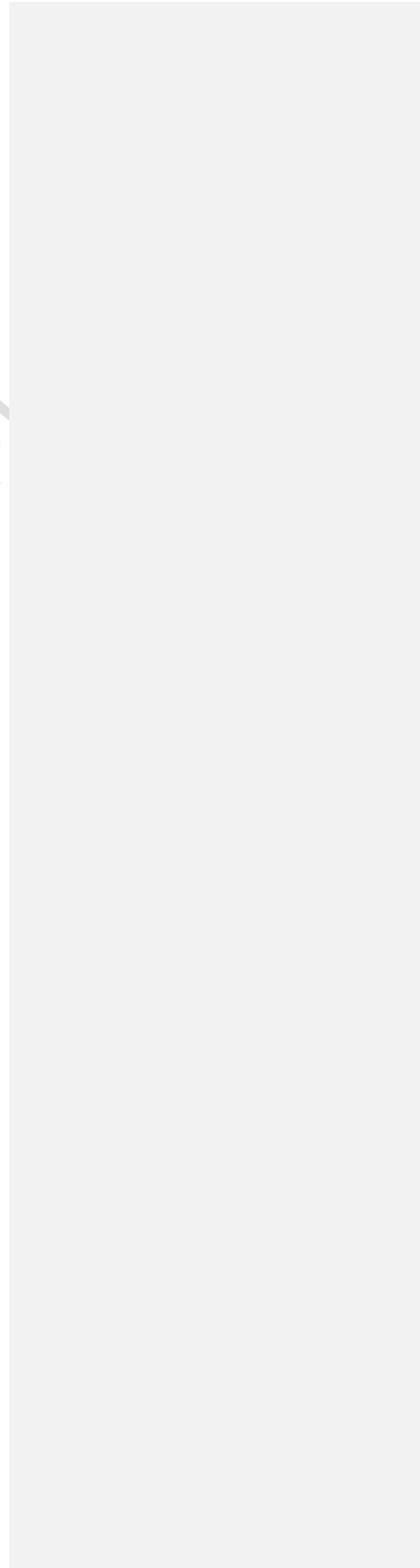


Table 7 Commercialized film forming system

Product	Drug	Company	Formulation type
Lamisil Once® [61]	Terbinafine Hydrochloride	Novartis Consumer Health, Australasia, Pty Ltd	Film forming Solution
Axiron® [62]	Testosterone	Lilly USA, LLC	Film forming spray
Medspray® the Patch-in- a-Can® [63]	Terbinafine Hydrochloride	MedPharm Ltd, UK	Film forming spray
Liqui-Patch technology [64]	Testosterone Hydrocortisone	Epinamics GmbH, Germany	Film forming spray
Durapeel Technology [65]	Ropivacane	Crescita Therapeutics.Inc	Film forming gel
PharmaDur®Technology [66]	Hydroquinone	Polytherapeutics, Inc	Film forming emulsion-gel